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O.M.R. Serial No.

प्रश्नपुस्तिका क्रमांक Question Booklet No.

प्रश्नपुस्तिका सीरीज Question Booklet Series

## M.Sc (Biotechnology) First Semester, Examination, February/March-2022 MBT-1002 General Biochemistry

Time: 1:30 Hours Maximum Marks-100

जब तक कहा न जाय, इस प्रश्नपुस्तिका को न खोलें

- निर्देश: 1. परीक्षार्थी अपने अनुक्रमांक, विषय एवं प्रश्नपुस्तिका की सीरीज का विवरण यथास्थान सही— सही भरें, अन्यथा मृल्यांकन में किसी भी प्रकार की विसंगति की दशा में उसकी जिम्मेदारी स्वयं परीक्षार्थी की होगी।
  - 2. इस प्रश्नपुस्तिका में 100 प्रश्न हैं, जिनमें से केवल 75 प्रश्नों के उत्तर परीक्षार्थियों द्वारा दिये जाने है। प्रत्येक प्रश्न के चार वैकल्पिक उत्तर प्रश्न के नीचे दिये गये हैं। इन चारों में से केवल एक ही उत्तर सही है। जिस उत्तर को आप सही या सबसे उचित समझते हैं, अपने उत्तर पत्रक (O.M.R. ANSWER SHEET)में उसके अक्षर वाले वृत्त को काले या नीले बाल प्वांइट पेन से पूरा भर दें। यदि किसी परीक्षार्थी द्वारा निर्धारित प्रश्नों से अधिक प्रश्नों के उत्तर दिये जाते हैं तो उसके द्वारा हल किये गये प्रथमतः यथा निर्दिष्ट प्रश्नोत्तरों का ही मूल्यांकन किया जायेगा।
  - 3. प्रत्येक प्रश्न के अंक समान हैं। आप के जितने उत्तर सही होंगे, उन्हीं के अनुसार अंक प्रदान किये जायेंगे।
  - 4. सभी उत्तर केवल ओ०एम०आर० उत्तर पत्रक (O.M.R. ANSWER SHEET) पर ही दिये जाने हैं। उत्तर पत्रक में निर्धारित स्थान के अलावा अन्यत्र कहीं पर दिया गया उत्तर मान्य नहीं होगा।
  - 5. ओ॰एम॰आर॰ उत्तर पत्रक (O.M.R. ANSWER SHEET) पर कुछ भी लिखने से पूर्व उसमें दिये गये सभी अनुदेशों को सावधानीपूर्वक पढ़ लिया जाय।
  - 6. परीक्षा समाप्ति के उपरान्त परीक्षार्थी कक्ष निरीक्षक को अपनी प्रश्नपुस्तिका बुकलेट एवं ओ०एम०आर० शीट पृथक-पृथक उपलब्ध कराने के बाद ही परीक्षा कक्ष से प्रस्थान करें।
  - 7. निगेटिव मार्किंग नहीं है।
- महत्वपूर्ण : प्रश्नपुस्तिका खोलने पर प्रथमतः जॉच कर देख लें कि प्रश्नपुस्तिका के सभी पृष्ठ भलीभाँति छपे हुए हैं। यदि प्रश्नपुस्तिका में कोई कमी हो, तो कक्ष निरीक्षक को दिखाकर उसी सीरीज की दूसरी प्रश्नपुस्तिका प्राप्त कर लें।

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1.	Aron	natic amino acids can be detected by:
	(A)	Sakaguchi reaction
	(B)	Millon-Nasse reaction
	(C)	Hopkins-Cole reaction
	(D)	Xanthoproteic reaction
2.	A wa	ter soluble vitamin deficient in egg is:
	(A)	Thiamin
	(B)	Ribofalvin
	(C)	Ascrobic acid
	(D)	Cobalamin
3.	A tra	ce element deficient in the milk is:
	(A)	Magnesium
	(B)	Copper
	(C)	Zinc
	(D)	Chloride
4.	The 1	imiting amino acid of rice is:
	(A)	Lysine
	(B)	Tryptophan
	(C)	Phenylalanine
	(D)	Tyrosine
5.	Dena	turation of proteins results in:
	(A)	Disruption of primary structure
	(B)	Breakdown of peptide bonds
	(C)	Destruction of hydrogen bonds
	(D)	Irreversible changes in the molecule

6.	In a protein molecule the disulphide bond is not broken by:
	(A) Reduction
	(B) Oxidation
	(C) Denaturation
	(D) X-ray diffraction
7.	Both $\alpha$ - helix and $\beta$ -pleated sheet conformation of proteins were proposed by:
	(A) Watson and Crick
	(B) Pauling and Corey
	(C) Waugh and King
	(D) Y.S. Rao
8.	The a-helix of proteins is:
	(A) A pleated structure
	(B) Made periodic by disulphide bridges
	(C) A non-periodic structure
	(D) Stabilised by hydrogen bonds between NH and CO groups of the main chain:
9.	Each turn of $\alpha$ -helix contains the amino acid residues (number):
	(A) 3.6
	(B) 3.0
	(C) 4.2
	(D) 4.5
10.	The amino acid form which synthesis of the protein of hair keratin takes place is:
	(A) Alanine
	(B) Methionine
	(C) Proline
	(D) Hydroxyproline

11.	An example of phosphoprotein present in egg yolk is:
	(A) Ovoalbumin
	(B) Ovoglobulin
	(C) Ovovitellin
	(D) Avidin
12.	An example of metalloprotein is:
	(A) Casein
	(B) Ceruloplasmin
	(C) Gelatin
	(D) Salmine
13.	The most of the ultraviolet absorption of proteins above 240 nm is due to their
	content of:
	(A) Tryptophan
	(B) Aspartate
	(C) Glutamate
	(D) Alanine
14.	Biuret reaction is specific for:
	(A) -CONH-linkages
	(B) -CSNH2 group
	(C) -(NH) NH2 group
	(D) All of these
15.	A ketogenic amino acid is:
	(A) Valine
	(B) Cysteine
	(C) Leucine
	(D) Threonine

16.	Amino acid with side chain containing basic groups is:
	(A) 2-Amino 5-guanidovaleric acid
	(B) 2-pyrrolidine carboxylic acid
	(C) 2-Amino 3-mercaptopropanoic acid
	(D) 2-Amino propanoic acid
17.	Since the pK values for aspartic acid are 2.0, 3.9 and 10.0, it follows that the
	isoelectric (pH) is:
	(A) 3.0
	(B) 3.9
	(C) 5.9
	(D) 6.0
18.	At neutral pH, a mixture of amino acids in solution would be predominantly:
	(A) Dipolar ions
	(B) Nonpolar molecules
	(C) Positive and monovalent
	(D) Hydrophobic
19.	Glycogen is present in all body tissues except:
	(A) Liver
	(B) Brain
	(C) Kidney
	(D) Stomach
20.	The normal resting state of humans, most of the blood glucose burnt as "fuel" is
	consumed by:
	(A) Liver
	(B) Brain
	(C) Kidneys
	(D) Adipose tissue

21.	Cerebrosides consist of mostly of this sugar:
	(A) Glucose
	(B) Fructose
	(C) Galactose
	(D) Arabinose
22.	The component of cartilage and cornea is:
	(A) Keratosulphate
	(B) Chondroitin sulphate
	(C) Cadmium sulphate
	(D) Antimony sulphate
23.	N-acetylneuraminic acid is an example of:
	(A) Sialic acid
	(B) Mucic acid
	(C) Glucuronic acid
	(D) Hippuric acid
24.	Honey contains the hydrolytic product of:
	(A) Lactose
	(B) Maltose
	(C) Inulin
	(D) Starch
25.	Tautomerisation is:
	(A) Shift of hydrogen
	(B) Shift of carbon
	(C) Shift of both
	(D) None of these

26.	An I	L-isomer of monosaccharide formed in human body is:
	(A)	L-fructose
	(B)	L-Erythrose
	(C)	L-Xylose
	(D)	L-Xylulose
27.	A po	ositive Benedict's test is not given by:
	(A)	Sucrose
	(B)	Lactose
	(C)	Maltose
	(D)	Glucose
28.	Rep	eating units of hyaluronic acid are:
	(A)	N-acetyl glucosamine and D-glucuronic acid (B) N-acetyl galactosamine and
		D-glucuronic acid
	(B)	N-acetyl glucosamine and galactose
	(C)	N-acetyl galactosamine and L-iduronic acid
	(D)	None of these
29.	The	glycosaminoglycan which does not contain uronic acid is:
	(A)	Dermatan sulphate
	(B)	Chondroitin sulphate
	(C)	Keratan sulphate
	(D)	Heparan sulphate
30.	Whi	ch of the following is a heteroglycan?
	(A)	Dextrins
	(B)	Agar
	(C)	Inulin
	(D)	Chitin

31.	The polysaccharide used in assessing the glomerular filtration rate (GFR) is:
	(A) Glycogen
	(B) Agar
	(C) Inulin
	(D) Hyaluronic acid
32.	The homopolysaccharide used for intravenous infusion as plasma substitute is:
	(A) Agar
	(B) Inulin
	(C) Pectin
	(D) Starch
33.	A dissaccharide formed by 1,1- glycosidic linkage between their monosaccharide
	units is:
	(A) Lactose
	(B) Maltose
	(C) Trehalose
	(D) Sucrose
34.	Invert sugar is:
	(A) Lactose
	(B) Sucrose
	(C) Hydrolytic products of sucrose
	(D) Fructose
35.	A sugar alcohol is:
	(A) Mannitol
	(B) Trehalose
	(C) Xylulose
	(D) Arabinose

36.	In glucose the orientation of the -H and -OH groups around the carbon atom 5
	adjacent to the terminal primary alcohol carbon determines:
	(A) D or L series
	(B) Dextro or levorotatory
	(C) $\alpha$ and $\beta$ anomers
	(D) Epimers
37.	Isomers differing as a result of variations in configuration of the -OH and -H on
	carbon atoms 2,3 and 4 of glucose are known as:
	(A) Epimers
	(B) Anomers
	(C) Optical isomers
	(D) Steroisomers
38.	The number of isomers of glucose is:
	(A) 2
	(B) 4
	(C) 8
	(D) 16
39.	The pentose sugar present mainly in the heart muscle is:
	(A) Lyxose
	(B) Ribose
	(C) Arabinose
	(D) Xylose
40.	The general formula of polysaccharides is:
	(A) $(C_6H_{10}O_5)n$
	(B) $(C_6H_{12}O_5)n$
	(C) $(C_6H_{10}O_6)n$
	(D) $(C_6H_{12}O_6)n$

41.	Genetic information of nuclear DNA is transmitted to the site of protein synthesis
	by:
	(A) rRNA
	(B) mRNA
	(C) tRNA
	(D) Polysomes
42.	The following substances are cell inclusions except:
	(A) Melanin
	(B) Glycogen
	(C) Lipids
	(D) Centrosome
43.	A lipid bilayer is permeable to:
	(A) Urea
	(B) Fructose
	(C) Glucose
	(D) Potassium
44.	The most active site of protein synthesis is the:
	(A) Nucleus
	(B) Ribosome
	(C) Mitochondrion
	(D) Cell sap
45.	The surface tension in intestinal lumen between fat droplets and aqueous medium is
	decreased by:
	(A) Bile Salts
	(B) Bile acids
	(C) Conc. H <sub>2</sub> SO <sub>4</sub>
	(D) Acetic acid

46.	The exchange of material takes place:
	(A) Only by diffusion
	(B) Only by active transport
	(C) Only by pinocytosis
	(D) All of these
47.	The degradative processes are categorized under the heading of:
	(A) Anabolism
	(B) Catabolism
	(C) Metabolism
	(D) None of the above
48.	Which of the following phospholipids is localized to a greater extent in the outer
	leaflet of the membrane lipid bilayer?
	(A) Choline phosphoglycerides
	(B) Ethanolamine phosphoglycerides
	(C) Inositol phosphoglycerides
	(D) Serine phosphoglycerides
49.	HDL is synthesized and secreted form:
	(A) Pancreas
	(B) Liver
	(C) Kidney
	(D) Muscle
50.	Synthesis of prostaglandinsis is inhibited by:
	(A) Aspirin
	(B) Arsenic
	(C) Fluoride
	(D) Cyanide

51.	Melting temperature of DNA is increased by its:
	(A) A and T content
	(B) G and C content
	(C) Sugar content
	(D) Phosphate content
52.	Ribozymes are:
	(A) Enzymes present in ribosomes
	(B) Enzymes which combine the ribosomal subunits
	(C) Enzymes which dissociate
	(D) Enzymes made up of RNA
53.	Nuclear DNA is present in combination with:
	(A) Histones
	(B) None-histones
	(C) Both (A) and (B)
	(D) None of these
54.	Cyclic AMP cab be formed from:
	(A) AMP
	(B) ADP
	(C) ATP
	(D) All of these
55.	Genetic code is:
	(A) Collection of codon
	(B) Collection of amino acids
	(C) Collection of purine nucleotide
	(D) Collection of pyrimidine nucleotide

56.	Phospholipase C is activated by:
	(A) Gs proteins
	(B) Gi proteins
	(C) Gq proteins
	(D) G12 proteins
57.	In the insulin molecule, the number of interchain disulphide brides is:
	(A) 1
	(B) 2
	(C) 3
	(D) 4
58.	Sodium is involved in the active uptake of:
	(A) D-Glucose
	(B) D-Galactose
	(C) L-Amino acids
	(D) All of these
59.	The major calcium salt in bones is:
	(A) Calcium carbonate
	(B) Calcium chloride
	(C) Calcium hydroxide
	(D) Calcium phosphate
60.	Superoxide radicals can be detoxified by:
	(A) Cytochrome c
	(B) Superoxide dismutase
	(C) Both (A) and (B)
	(D) None of these

61.	Fluo	rosis occurs due to:
	(A)	Drinking water containing less than 0.2 ppm of fluorine
	(B)	Drinking water containing high calcium
	(C)	Drinking water containing greater than 1.2 ppm of fluorine
	(D)	Drinking water containing heavy metals
62.	Iodii	ne is stored in:
	(A)	Thyroid gland as thyroglobulin
	(B)	Liver
	(C)	Intestine
	(D)	Skin
63.	Exce	ess intake of cobalt for longer periods leads to:
	(A)	Polycythemia
	(B)	Megaloblastic anemia
	(C)	pernicious anemia
	(D)	Microcytic anemia
64.	A no	onspecific intracellular antioxidant is:
	(A)	Chromium
	(B)	Magnesium
	(C)	Selenium
	(D)	Nickel
65.	The	richest source of copper is:
	(A)	Liver
	(B)	Milk
	(C)	Legumes
	(D)	Green leafy vegetables

66.	Daily	y requirement of iron for normal adult male is about:
	(A)	5 mg
	(B)	10 mg
	(C)	15 mg
	(D)	20 mg
67.	Calc	ium is excreted by:
	(A)	Kidney
	(B)	Kidney and intestine
	(C)	Kidney and liver
	(D)	Kidney and pancreas
68.	Folic	e acid or folate consists of the:
	(A)	Base pteridine, p-amino benzoic acid and asparate
	(B)	Base Purine, p-amino benzoic acid and glutamate
	(C)	Base pteridine, p-amino benzoic acid and glutamate
	(D)	Base purine, p-hydroxy benzoic acid and glutamate
69.	Cons	sumption of raw eggs can cause deficiency of:
	(A)	Biotin
	(B)	Pantothenic acid
	(C)	Riboflavin
	(D)	Thiamin
70.	Vitai	min B6 deficiency may occur during therapy with:
	(A)	Isoniazid
	(B)	Terramycin
	(C)	Sulpha drugs
	(D)	Aspirin

71.	The vitamin which does not contain a ring in the structure is:
	(A) Pantothenic acid
	(B) Vitamin D
	(C) Riboflavin
	(D) Thiamin
72.	Sterilised milk lacks in:
	(A) Vitamin A
	(B) Vitamin D
	(C) Vitamin C
	(D) Thiamin
73.	In human body highest concentration of ascorbic acid is found in:
	(A) Liver
	(B) Adrenal cortex
	(C) Adrenal medulla
	(D) Spleen
74.	The most important natural antioxidant is:
	(A) Vitamin D
	(B) Vitamin E
	(C) Vitamin B <sub>12</sub>
	(D) Vitamin K
75.	Vitamin synthesized by bacterial in the intestine is:
	(A) A
	(B) C
	(C) D
	(D) K

76.	Vitamin K is found in:
	(A) Green leafy plants
	(B) Meat
	(C) Fish
	(D) Milk
77.	Function of Vitamin A:
	(A) Healing epithelial tissues
	(B) Protein synthesis regulation
	(C) Cell growth
	(D) All of these
78.	The cyclic ring present in all the steroids:
	(A) Cyclopentano perhydrophenanthrene
	(B) Nitropentano
	(C) Both (A) and (B)
	(D) None of these
79.	Leukotrienes cause:
	(A) Increase in capillary permeability
	(B) Aggregation of platelets
	(C) Bronchodilatation
	(D) None of these
80.	Sphingomyelins:
	(A) Phospholipids
	(B) Nitrolipids
	(C) Alcohols
	(D) None of these

81.	Saliv	vary lipase hydrolyses the ester bond at:
	(A)	Position 1 of triglycerides
	(B)	Position 2 of triglycerides
	(C)	Position 3 of triglycerides
	(D)	All of these
82.	Lyso	plecithin is formed form lecithin by removal of:
	(A)	Fatty acid from position 1
	(B)	Fatty acid from position 2
	(C)	Phosphorylcholine
	(D)	Choline
83.	Thro	omboxanes cause:
	(A)	Vasodilation
	(B)	Bronchoconstriction
	(C)	Platelet aggregation
	(D)	All of these
84.	Cho	lesterol is transported from liver to extrahepatic tissues by:
	(A)	Chylomicrons
	(B)	VLDL
	(C)	HDL
	(D)	LDL
85.	The	major storage form of lipids is:
	(A)	Esterified cholesterol
	(B)	Glycerophospholipids
	(C)	Triglycerides
	(D)	Sphingolipids

86.	The	major lipid in chylomicrons is:	
	(A)	Triglycerides	
	(B)	Phospholipids	
	(C)	Cholesterol	
	(D)	Free fatty acids	
87.	Triglycerides are:		
	(A)	Heavier than water	
	(B)	Major constituents of membranes	
	(C)	Non-polar	
	(D)	Hydrophilic	
88.	The	end products of saponification:	
	(A)	Glycerol	
	(B)	Acid	
	(C)	Soap	
	(D)	Both (A) and (C)	
89.	Athe	erosclerosis and coronary heart diseases are associated with the diet:	
	(A)	High in total fat and saturated fat	
	(B)	Low in protein	
	(C)	High in protein	
	(D)	High in carbohydrate	
90.	Dieta	ary fibre denotes:	
	(A)	Undigested proteins	
	(B)	Plant cell components that cannot be digested by own enzymes	
	(C)	All plant cell wall components	
	(D)	All non-digestible water insoluble polysaccharide	

91.	Mol	ecular formula of cholesterol is:
	(A)	$C_{27}H_{45}OH$
	(B)	$C_{29}H_{47}OH$
	(C)	$C_{29}H_{46}OH$
	(D)	$C_{23}H_{41}OH$
92.	Dete	erioration of food (rancidity) is due to presence of:
	(A)	Cholesterol
	(B)	Vitamin E
	(C)	Peroxidation of lipids
	(D)	Phenolic compounds
93.	Phos	spholipid acting as surfactant is:
	(A)	Cephalin
	(B)	Phosphatidyl inositol
	(C)	Lecithin
	(D)	Phosphatidyl serine
94.	The	importance of phospholipids as constituent of cell membrane is because they
	poss	ess
	(A)	Fatty acids
	(B)	Both polar and nonpolar groups
	(C)	Glycerol
	(D)	Phosphoric acid
95.	The	number of double bonds in arachidonic acid is:
	(A)	1
	(B)	2
	(C)	4
	(D)	6

96.	A fa	tty acid which is not synthesized in the body and has to be supplied in the diet
	is:	
	(A)	Palmitic acid
	(B)	Lauric acid
	(C)	Linolenic acid
	` /	Plamitoleic acid
97.	A lipid containing alcoholic amine residue is:	
	_	Phosphatidic acid
	(B)	Ganglioside
	(C)	Glucocerebroside
	( )	
00	` /	Sphingomyelin
98.		-Protein amino acids are:
	(A)	Ornithine
	(B)	$\beta$ -alanine
	(C)	Y-amino butyric acid
	(D)	All of these
99.	All of the following statements about aspartate are true except:	
	(A)	It is non-essential amino acid
	(B)	It is a dicarboxylic amino acid
	(C)	It can be synthesized from pyruvate and glutamate
	(D)	It can be converted into asparagine
100.	Edman's reagent contains:	
	(A)	Phenylisothiocyanate
	(B)	1-Fluoro-2, 4-dinitrobenzene
	(C)	Dansyl Chloride
	(D)	tBOC azide

## Rough Work / रफ कार्य

## DO NOT OPEN THE QUESTION BOOKLET UNTIL ASKED TO DO SO

- 1. Examinee should enter his / her roll number, subject and Question Booklet Series correctly in the O.M.R. sheet, the examinee will be responsible for the error he / she has made.
- 2. This Question Booklet contains 100 questions, out of which only 75 Question are to be Answered by the examinee. Every question has 4 options and only one of them is correct. The answer which seems correct to you, darken that option number in your Answer Booklet (O.M.R ANSWER SHEET) completely with black or blue ball point pen. If any examinee will mark more than one answer of a particular question, then the first most option will be considered valid.
- 3. Every question has same marks. Every question you attempt correctly, marks will be given according to that.
- 4. Every answer should be marked only on Answer Booklet (O.M.R ANSWER SHEET). Answer marked anywhere else other than the determined place will not be considered valid.
- 5. Please read all the instructions carefully before attempting anything on Answer Booklet(O.M.R ANSWER SHEET).
- 6. After completion of examination please hand over the Answer Booklet (O.M.R ANSWER SHEET) to the Examiner before leaving the examination room.
- 7. There is no negative marking.

**Note:** On opening the question booklet, first check that all the pages of the question booklet are printed properly in case there is an issue please ask the examiner to change the booklet of same series and get another one.