Roll No	 				Question Booklet	Number
O. M. R. Serial No.						

M. Sc. (Industrial Chemistry) (Fourth Semester) EXAMINATION, July, 2022

CHEMISTRY OF LIFE

Paper	Cod	e	
MSIC	4	0	3

Questions Booklet Series

B

[Maximum Marks : 100

Time: 1:30 Hours]

Instructions to the Examinee:

- 1. Do not open the booklet unless you are asked to do so.
- 2. The booklet contains 100 questions. Examinee is required to answer any 75 questions in the OMR Answer-Sheet provided and not in the question booklet. If more than 75 questions are attempted by student, then the first attempted 75 questions will be considered for evaluation. All questions carry equal marks.
- 3. Examine the Booklet and the OMR Answer-Sheet very carefully before you proceed. Faulty question booklet due to missing or duplicate pages/questions or having any other discrepancy should be got immediately replaced.

परीक्षार्थियों के लिए निर्देश :

- प्रश्न-पुस्तिका को तब तक न खोलें जब तक आपसे कहा न जाए।
- 2. प्रश्न-पुस्तिका में 100 प्रश्न हैं। परीक्षार्थी को किन्हीं 75 प्रश्नों को केवल दी गई OMR आन्सर-शीट पर ही हल करना है, प्रश्न-पुस्तिका पर नहीं। यदि छात्र द्वारा 75 से अधिक प्रश्नों को हल किया जाता है तो प्रारम्भिक हल किये हुए 75 उत्तरों को ही मूल्यांकन हेतु सम्मिलित किया जाएगा। सभी प्रश्नों के अंक समान हैं।
- उत्तर अंकित करने से पूर्व प्रश्न-पुस्तिका तथा OMR आन्सर-शीट को सावधानीपूर्वक देख लें। दोषपूर्ण प्रश्न-पुस्तिका जिसमें कुछ भाग छपने से छूट गए हों या प्रश्न एक से अधिक बार छप गए हों या उसमें किसी अन्य प्रकार की कमी हो, तो उसे तुरन्त बदल लें।

(शेष निर्देश अन्तिम पृष्ठ पर)

(Only for Rough Work)

- 1. Hydrophobic interaction generally contribute to the :
 - (A) folding and shaping of a protein
 - (B) defolding and shaping of a protein
 - (C) folding and non-shaping of a protein
 - (D) defolding and non-shaping of a protein
- 2. Heamoglobin has:
 - (A) Secondary structure of protein
 - (B) Tertiary structure of protein
 - (C) Quaternary structure of protein
 - (D) Both Secondary structure and tertiary structure
- 3. Which one of the following is responsible for metabolism of carbohydrate ?
 - (A) Insulin
 - (B) Myoglobin
 - (C) Hormones
 - (D) Fats
- 4. 1-fluoro-2, 4-dinitrobenzene (FDNB) is used for amino-end degradation in :
 - (A) Edman's method
 - (B) Dansyl method
 - (C) Sanger's method
 - (D) Enzymatic method

- 5. Which one of the following is used amino-end degradation in Dansyl method?
 - (A) 1-dimethylamino naphthalene-5-sulphonyl chloride
 - (B) 1-fluoro-2, 4-dinitrobenzene
 - (C) Phenyl isothiocyanate
 - (D) Lithium aluminium hydrie
- 6. Leucine amino peptidase enzyme attacks proteins only at :
 - (A) the end which contains free amino group
 - (B) the end which contains free carboxyl group
 - (C) the end which contain free amino group or carboxyl group
 - (D) the middle of the protein
- 7. In hydrazinolysis the peptide or protein is heated with:
 - (A) anhydrous hydrazine
 - (B) hydrous hydrazine
 - (C) Both (A) and (B)
 - (D) Aminoacid hydrazides

- 8. Pepsin attacks peptides having:
 - (A) NH part of leucine, aspartic acid and CO part of glysine, arginine
 - (B) CO part of leucine, aspartic acid and NH part of glycine, arginine
 - (C) Both (A) and (B)
 - (D) None of the above
- 9. Cyanogen bromide in aqueous formic acid attacks only those peptides in which:
 - (A) CO group of methionine residues
 - (B) CO group of non-methionine residue
 - (C) NH group of methionine residue
 - (D) NH group of non-methionine residue
- 10. Saponifiable lipids are hydrolysed by:
 - (A) heat, alkali or acid solution
 - (B) only by heat
 - (C) heat and alkali solution
 - (D) heat and acid solution
- 11. The proteins help to protect from any diseases in the body is :
 - (A) Enzymes
 - (B) Storage proteins
 - (C) Transport proteins
 - (D) Antibodies

- 12. Which protein is called transport protein?
 - (A) Haemoglobin
 - (B) Keratin
 - (C) Enzymes
 - (D) Oval bumin
- 13. In hair which protein is found?
 - (A) Myosin
 - (B) Elastin
 - (C) Keratin
 - (D) Tropocollage
- 14. Disulphide bonds are formed between:
 - (A) Cysteine residues that are close together.
 - (B) Histidine residue that are close together.
 - (C) Protein residue that are close together.
 - (D) Phenylalanine residue that are close together.
- 15. Primary structure of protein represents:
 - (A) Linear sequence of amino acids joined by peptide bond
 - (B) 3-dimensional structure of protein
 - (C) Helical structure of protein
 - (D) Subunit structure of protein

16.	The structure in which all peptide chains	19.	The action of certain is		
	are stritched out to full extension and laid		mediated through phosphatidyl inositol.		
	side by through intermolecular hydrogen		(A) enzymes		
	bond is called:		(B) hormones		
	(A) Tertiary structure		(C) proteins		
	(B) β -pleated sheet		(D) carbohydrate		
	(C) Quaternary structure	20.	Which one of the following is responsible		
	(D) α-helix		for deterioration of food (rancidity) as		
			well as for damage to tissues ?		
17.	Fibrous and globular proteins are		(A) Oxidation of lipid exposed to		
	classified on the basis of:		oxygen		
	(A) Primary structure		(B) Peroxidation of lipid exposed to		
	(B) Tertiary structure		oxygen		
	(C) Secondary structure		(C) Peroxidation of lipid exposed to		
	(D) Quaternary structure		CO_2		
			(D) Oxidation of lipid exposed to CO ₂		
18.	perform external protective	21.	Basic structure in biological membrane		
	function.		consists of :		
	(A) Waxes		(A) bilayer of amphiphatic lipid		
	(B) Alcohols		(B) single layer of amphiphatic lipid		
	(C) Phosphoglycerides		(C) bilayer of protein		
	(D) All of the above		(D) a layer of carbohydrate		

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Set-B

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22.	enzyme catlayze oxidation-	26.	An enzyme that joins the ends of two
	reduction reaction where electron(s) is/		strands of nucleic acid is:
	are transferred.		(A) Polymerase
	(A) Transferase		(B) Synthetase
	(B) Oxidoreduatase		(C) Transferase
	(C) Hydrolase		(D) Ligase
	(D) Isomerase		•
23.	Pepsin hydrolyzes in proteins.	27.	Diastase takes part in digestion of
	(A) Hydrogen bonds		(A) Starch
	(B) Peptide bonds		(B) Protein
	(C) Sulphide bonds		(C) Fat
	(D) Carbon-carbon double bond		(D) Amino acids
	(C = C)	28.	Enzyme catalysing rearrangement of
24.	The nature of enzyme is:		functional groups or atomic grouping
	(A) Vitamin		without altering molecular weight or
	(B) Lipid		number of atom is:
	(C) Carbohydrate		(A) Oxidoreductase
	(D) Protein		(B) Ligase
25.	The statement about enzymes is true:		(C) Isomerase
	(A) enzymes increases reactions by		(D) Hydrolase
	lowering the activation energy.	29.	Enzyme activity is highest when the
	(B) enzymes do not alter the overall	2).	substrate concentration is:
	change in free energy for a reaction		
	(C) enzymes are protein whose three-		(A) Small
	dimensional structure is key to their		(B) High
	function.		(C) Unlimited
	(D) All of the above		(D) All of the above

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Set-B

MSIC-403

30.	If the enzymeg amount is kept constant	34.	The catalytic efficiency of two distinct
	and the substrate is then gradually		enzymes can be compared based on
	, the reaction will increase until it		which of following factors?
	reaches a maximum.		(A) Size of the enzyme
	(A) decreased		(B) K_m
	(B) increased		(C) pH of optimum value
	(C) kept constant		(D) Product formation
	(D) None of the above	35.	Types of inhibition pattern based on
31.	The rate determining step of Michaelis-		Michaelis-Menten equation are:
	Menten kinetics is		(A) Competitive
	(A) The complex formation step		(B) Non-competitive
	(B) Complex dissociation step to		(C) Reversible
	produce products		(D) All of the above
	(C) Product formation step	36.	Which of the following steps is assumed
	(D) All of the above		to be the slowest step in the Michaelis-
32.	The molecule which acts directly on an		Menten equation ?
32.	enzyme to lower its catalytic rate is		(A) The substrate consuming step
	(A) Modulator		(B) Formation of enzyme substrate
	(B) Inhibitor		complex
	(C) Accelerator		(C) The product releasing step
	(D) None of the above		(D) None of the above
33.	The inhibitor molecule structurally and	37.	Lock and key theory is based on the compatibility of:
	chemically similar to the substrate		(A) enzyme and product
	is		(B) enzyme and substrate
	(A) non-competitive inhibitor		(C) enzyme substrate complex and
	(B) competitive inhibitor		product
	(C) Both (A) and (B)		(D) enzyme and enzyme substrate
	(D) None of the above		complex

complex

38.	Treatment of influenza via the neuro-	42.	Purine base found in RNA is:
	aminidase inhibitor (Relenza) is an		(A) Thymine
	example of:		(B) Uracil
	(A) Competitive inhibitor		(C) Guanine
	(B) Non-competitive inhibitor		(D) Cytocine
	(C) Reversible enzyme inhibitor	43.	What is the composition of nucleotide?
	•		(A) base-sugar
	(D) Irreversible enzyme inhibitor		(B) base-phosphate
39.	Irreversible enzyme inhibitors bind		(C) base-sugar-phosphate
	to the enzyme, thus they		(D) sugar-phosphate
	dissociate very closely from the enzyme.	44.	Group of adjacent nucleotides are joined
	(A) losely		at position.
	(B) tightly		(A) 3, 5
	(C) normally		(B) 1, 4
	•		(C) 2, 4
	(D) All of the above		(D) 2, 3
40.	Nucleosides contain:	45.	Which of the following base contains two
	(A) base-phosphate		keto groups ?
	(B) base-sugar		(A) adenine
	(C) sugar-phosphate		(B) thymine
	(D) base-sugar phosphate		(C) gaunine
	(D) base-sugai phosphate		(D) cytocine
41.	The sugar molecule present in nucleotide	46.	Michaelis constant K_m is the substrate
	is:		concentration at which rate of reaction is
			the maximal velocity attainable
	(A) hexose		at a particular concentration of enzyme.
	(B) pentose		(A) equal
	(C) tetrose		(B) half
			(C) double
	(D) glucose		(D) triple

47.	Identify the complementary strand of the	51.	Nucleic acids combine with which	
	DNA primary structure ATGCCGATC:		biomolecule?	
	(A) AUGCCGUAC		(A) Carbohydrates	
	(B) UACGGCUAG		(B) Lipids	
	(C) TACGGCTAG		(C) Proteins	
	(D) GATCGGCAT		(D) Amino acids	
48.	Which part of the nucleotide is	52.	Which of the following is pyramidine	
	responsible for the formation of bonds in		nucleotide?	
	DNA double helix ?		(A) Uracil	
	(A) base		(B) Cytocine	
	(B) sugar		(C) Thymine	
	(C) phosphate group		(D) All of the above	
	(D) –OH group		(D) All of the above	
49.	The number of hydrogen bond present	53.	Phosphate group is attached to which	
	between cytocine and guanine are:		carbon of sugar molecule?	
	(A) five		(A) C-5	
	(B) four		(B) C-4	
	(C) three		(C) C-3	
	(D) two		(D) C-1	
50.	The backbone sugar of DNA is:	54.	Which ratio is constant for DNA?	
	(A) ribose		(A) $A + U/G + C$	
	(B) deoxyribose		(B) $A + G/T + C$	
	(C) fructose		(C) $A + T/G = C$	
	(D) oxyribose		(D) $A = C/U + G$	

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	(D) cytocine		(D) All of the above
	(C) thymine		(C) Phosphatidyl inositol
	(B) uracil		(B) Cephalin
	(A) adenine		(A) Lecithin
	in the following:		in emulsification of fat.
58.	Identify the purine base of nucleic acids	62.	is the surface active agent and help
50	Identify the numine been of nucleic saids		(D) All of the above
	(D) phosphoric acid		environment
	(C) amino acid		(C) favourable and unfavourable
	(B) nitrogen base		(B) favourable environment
	(A) pentose sugar		(A) an unfavourable environment
	hydrolysis of RNA?		are exposed to :
57.	Which of the following does not yield on	61.	Only the ends or edges of bilayer sheet
	(D) None of the above		(D) None of the above
	(D) None of the above		(C) Both (A) and (B)
	(C) G+C		(B) Micelles
	(B) A + T		(A) Liposomes
	(A) $A+G$		least one lipid bilayer.
	ds-DNA may be due to high content of:	60.	is a spherical vesicle having at
56.	An increase melting temperature for a		(D) oxo-linkage of sugar
	(D) nucleoside		(C) peptide
	(C) nucleotide		(B) H-bond
	(B) vitamin		(A) phosphodiester
	(A) nucleic acid		structure ?
	(A) 1: '1		stabilizes the DNA double strand

59. Which among the following bonds

55. ATP is a:

	deriv	atives of :		fatty	acid?
	(A)	Phophatidyl glycerol		(A)	Linolenic acid
	(B)	Cardiolipin		(B)	Palmitic acid
	(C)	Sphingosine Name of the allower		, ,	Linoleic acid
	(D)	None of the above		(C)	
64.	•	acids are linked to before		(D)	Both (A) and (B)
		are oxidised in lipid metabolism.	69.	Whi	ch of the following undergoes β -
	(A)	enzyme		oxid	ation ?
	(B)	cofactor			
	(C) (D)	coenzyme-A All of the above		(A)	Saturated fatty acids
	(D)	All of the above		(B)	Monounsaturated fatty acids
65.		activation reaction of fatty acid		(C)	Polyunsaturated fatty acids
		rs on the :		(D)	All of the above
	(A)	Mitochondrial membrane	70	TD1	
	(B)	Cell membrane	70.	The	long-chain fatty acids get transported
	(C)	Nuclear membrane		throu	igh the inner mitochondrial
	(D)	Golgi complex		mem	abrane:
66.	Oxid	ation of fatty acid produces large		(A)	freely
	quan	tity of:		(B)	as cornitine derivative
	(A)	ADP (Adenosine diphosphate)		, ,	
	(B)	Adensosine triphosphate (ATP)		(C)	as acyl-CoA derivative
	(C)	Adenosine monophosphate (AMP)		(D)	require sodium-dependent carrier
	(D)	None of the above	71.	Whi	ch of the following product is
67.	In wl	hich part of the cell the enzymes for		relea	sed in α -oxidation of fatty acids ?
	β -ox	aidation is present?			CoA
	(A)	Golgi apparatus		(A)	
	(B)	Nucleus		(B)	H_2O
	(C)	Cytosol		(C)	CO_2
	(D)	Mitochondria		(D)	Acetyl CoA
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68.

Which one the following is an essential

Sphingolipids or Shingophospholipids are

63.

- 72. Which of the following factors is not responsible for the denaturation of proteins?
 - (A) pH change
 - (B) Heat
 - (C) Charge
 - (D) Organic solvents
- 73. What type of bond is present between the amino acid?
 - (A) Acidic bond
 - (B) Ionic bond
 - (C) Peptide bond
 - (D) Coordinate bond
- 74. Which of the following cell organelles is involved in the process of protein synthesis?
 - (A) Vesicles
 - (B) Mitochondria
 - (C) Ribosomes
 - (D) Vacuoles
- 75. Which of the following is false about fibrous proteins?
 - (A) Keratin and collagen are the best examples.
 - (B) It is in rod or wire like shape.
 - (C) Hemoglobin is the best example.
 - (D) It provides structural support for cells and tissues.

- 76. Smooth endoplasmic reticulum:
 - (A) actively participate in protein synthesis
 - (B) does not actively participate in protein synthesis
 - (C) participate in protein and lipid synthesis
 - (D) does not participate in lipid and protein synthesis
- 77. Which one of the following is formed when cell feeds on the intracellular organelles such as mitochondria?
 - (A) Autophagie vacuoles
 - (B) Residual bodies
 - (C) Secondary lysosomes
 - (D) All of the above
- 78. Which one is the fundamental and structural unit of all living organisms?
 - (A) tissue
 - (B) organs
 - (C) cell
 - (D) organ system

79.	are directly involved in normal	83.	Which of the following cell organelle is
	growth, development and reproduction of		responsible for transporting modifying
	living organism.		and packaging proteins and lipids?
	(A) Secondary metabolites		(A) Endoplasmic reticulum
	(B) Primary metabolites		(B) Golgi complex
	(C) Both (A) and (B)		(C) Ribosome
	(D) None of the above		(D) Basal granules
80.	Cell biology is the	84.	is a semi-rigid, laminated,
	(A) Study of metaphase of a cell		external and non-living covering of cell.
	(B) Study of cell division only		(A) Plasma membrane
	(C) Study of cancereous cell		(B) Cell wall
	(D) Study of cell structure and		(C) Cytoplasm
	functions		(D) Nucleoplasm
81.	Which of the following is known as	85.	Membrane of endoplasmic reticulum is
	power house of the cell?		rough due to:
	(A) Cytoplasm		(A) absence of ribosomes
	(B) Lysosome		(B) presence of ribosomes
	(C) Mitochondria		(C) Both (A) and (B)
	(D) Nuclei		(D) None of the above
82.	Which of the following is known as the	86.	Smooth endoplasmic reticulum is found
	suicide bag of a cell?		in:
	(A) Golgi complex		(A) Liver cells
	(B) Lysosome		(B) Pancreatic cells
	(C) Endoplasmic reticulum		(C) Adipose cells
	(D) Ribosome		(D) Plasma cells

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Set-B

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87.	Glyoxysome present in plant cells	91.	Space between nuclear envelope and
	contains enzymes for :		nuclear membrane is filled by:
	(A) Fatty acid metabolism		(A) Cytoplasm
	(B) Glyconeogenesis		(B) Cytoplasmic matrix
	(C) Both (A) and (B)		(C) Nucleoplasm
	(D) None of the above		(D) Nucleolus
88.	The actual respiratory organs of the cells where the food stuffs i. e. carbohydrates	92.	The proteins which take basic strain are:
	and fats are completely oxidised into CO ₂		(A) Nucleoprotamines and
	and H ₂ O is:		Nucleohistones
	(A) Golgi bodies		(B) Histones with rich lysine
	(B) Mitochondria		(C) Histone with rich arginine
	(C) Vacuoles		(D) Non-histone protein
	(D) Ribosomes		(b) Non-instone protein
89.	ribosomes occur in eukaryotic	93.	Basic unit of protein is:
	cells of plants and animals.		(A) Peptides
	(A) 80S (B) 70S		(B) Amino acids
	(B) 70S (C) 65S		(C) Enzymes
	(D) 90S		(D) All of the above
90.	Nuclear membrane is:	94.	Which one of the following on hydrolysis
	(A) bounded by 2 membranes of		yield non-proteineous substances and
	lipoprotein (B) bounded by 2 membranes of lipid		amino acid?
	only		(A) Conjugated protein
	(C) bounded by 2 membranes of carbohydrate		(B) Derived protein
	(D) bounded by 2 membranes of lipid		(C) Simple protein
	and protein		(D) Secondary protein

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	(D)	Respiratory protein		(D)	Hydrogen and hydrophobic
	(C)	Transport protein		(C)	Hydrogen and ionic
	(B)	Contractite protein		, ,	
	(A)	Storage protein		(B)	Hydrogen bond only
	of:			(A)	Hydrogen, ionic and hydrophobic
97.	Seru	m albumin comes under the category		struc	ture of proteins are :
	(D)	None of the above	100.	Bono	ls responsible for the 3-dimensional
	(C)	Intermediate protein		(D)	All of the above
	(B)	Fibrous protein		(D)	All of the above
	(A)	Globular protein		(C)	oxygen
	acid	?		(B)	carbon dioxide
	cond	ensation product of neutral amino		(A)	water
96.	Whic	ch one of the following is the linear	99.	Myo	globin carries in muscles.
		of heat and chemicals on protein		(D)	Covalent boliding
	(D)	Insoluble proteins formed by action		(D)	Covalent bonding
		of chemicals on protein			bonding
	(C)	Soluble proteins formed by action		(C)	Inter and intramolecular Hydrogen
	, ,	of a heat on protein		(B)	Intramolecular Hydrogen bonding
	(B)	Soluble proteins formed by action		(A)	Intermolecular Hydrogen bonding
	()	of heat on protein		prote	in is stabilized by .
	(A)	Insoluble proteins formed by action		prote	in is stabilized by:

98. The α -helix secondary structure of

95. Denatured proteins are:

4. Four alternative answers are mentioned for each question as—A, B, C & D in the booklet. The candidate has to choose the most correct/appropriate answer and mark the same in the OMR Answer-Sheet as per the direction:

Example:

Question:

Q.1 (A) (C) (D)
Q.2 (A) (B) (C) (D)
Q.3 (A) (C) (D)

Illegible answers with cutting and over-writing or half filled circle will be cancelled.

- 5. Each question carries equal marks. Marks will be awarded according to the number of correct answers you have.
- 6. All answers are to be given on OMR Answer sheet only. Answers given anywhere other than the place specified in the answer sheet will not be considered valid.
- 7. Before writing anything on the OMR Answer Sheet, all the instructions given in it should be read carefully.
- 8. After the completion of the examination candidates should leave the examination hall only after providing their OMR Answer Sheet to the invigilator. Candidate can carry their Question Booklet.
- 9. There will be no negative marking.
- 10. Rough work, if any, should be done on the blank pages provided for the purpose in the booklet.
- 11. To bring and use of log-book, calculator, pager and cellular phone in examination hall is prohibited.
- 12. In case of any difference found in English and Hindi version of the question, the English version of the question will be held authentic.
- Impt.: On opening the question booklet, first check that all the pages of the question booklet are printed properly. If there is ny discrepancy in the question Booklet, then after showing it to the invigilator, get another question Booklet of the same series.

4. प्रश्न-पुस्तिका में प्रत्येक प्रश्न के चार सम्भावित उत्तर—
A, B, C एवं D हैं। परीक्षार्थी को उन चारों विकल्पों में से
एक सबसे सही अथवा सबसे उपयुक्त उत्तर छाँटना है।
उत्तर को OMR आन्सर-शीट में सम्बन्धित प्रश्न संख्या में
निम्न प्रकार भरना है:

उदाहरण :

प्रश्न :

प्रश्न 1 (A) (C) (D) प्रश्न 2 (A) (B) (D) प्रश्न 3 (A) (C) (D)

अपठनीय उत्तर या ऐसे उत्तर जिन्हें काटा या बदला गया है, या गोले में आधा भरकर दिया गया, उन्हें निरस्त कर दिया जाएगा।

- 5. प्रत्येक प्रश्न के अंक समान हैं। आपके जितने उत्तर सही होंगे, उन्हीं के अनुसार अंक प्रदान किये जायेंगे।
- 6. सभी उत्तर केवल ओ. एम. आर. उत्तर-पत्रक (OMR Answer Sheet) पर ही दिये जाने हैं। उत्तर-पत्रक में निर्धारित स्थान के अलावा अन्यत्र कहीं पर दिया गया उत्तर मान्य नहीं होगा।
- 7. ओ. एम. आर. उत्तर-पत्रक (OMR Answer Sheet) पर कुछ भी लिखने से पूर्व उसमें दिये गये सभी अनुदेशों को सावधानीपूर्वक पढ लिया जाये।
- 8. परीक्षा समाप्ति के उपरान्त परीक्षार्थी कक्ष निरीक्षक को अपनी OMR Answer Sheet उपलब्ध कराने के बाद ही परीक्षा कक्ष से प्रस्थान करें। परीक्षार्थी अपने साथ प्रश्न-पुस्तिका ले जा सकते हैं।
- 9. निगेटिव मार्किंग नहीं है।
- 10. कोई भी रफ कार्य, प्रश्न-पुस्तिका के अन्त में, रफ-कार्य के लिए दिए खाली पेज पर ही किया जाना चाहिए।
- 11. परीक्षा-कक्ष में लॉग-बुक, कैलकुलेटर, पेजर तथा सेल्युलर फोन ले जाना तथा उसका उपयोग करना वर्जित है।
- 12. प्रश्न के हिन्दी एवं अंग्रेजी रूपान्तरण में भिन्नता होने की दशा में प्रश्न का अंग्रेजी रूपान्तरण ही मान्य होगा।

महत्वपूर्ण : प्रश्नपुस्तिका खोलने पर प्रथमतः जाँच कर देख लें कि प्रश्न-पुस्तिका के सभी पृष्ठ भलीभाँति छपे हुए हैं। यदि प्रश्नपुस्तिका में कोई कमी हो, तो कक्षनिरीक्षक को दिखाकर उसी सिरीज की दूसरी प्रश्न-पुस्तिका प्राप्त कर लें।