Roll No	 				Question Booklet Number
O. M. R. Serial No.					

M. Sc. (Biotechnology) (Second Semester) EXAMINATION, July, 2022

PHYSIOLOGY & METABOLISM

Paper Code				
MBT	2	0	0	3

Questions Booklet Series

A

[Maximum Marks : 100

Time: 1:30 Hours]

Instructions to the Examinee:

- 1. Do not open the booklet unless you are asked to do so.
- 2. The booklet contains 60 questions. Examinee is required to answer any 50 questions in the OMR Answer-Sheet provided and not in the question booklet. If more than 50 questions are attempted by student, then the first attempted 50 questions will be considered for evaluation. All questions carry equal marks.
- 3. Examine the Booklet and the OMR Answer-Sheet very carefully before you proceed. Faulty question booklet due to missing or duplicate pages/questions or having any other discrepancy should be got immediately replaced.

परीक्षार्थियों के लिए निर्देश :

- प्रश्न-पुस्तिका को तब तक न खोलें जब तक आपसे कहा न जाए।
- 2. प्रश्न-पुस्तिका में 60 प्रश्न हैं। परीक्षार्थी को किन्हीं 50 प्रश्नों को केवल दी गई OMR आन्सर-शीट पर ही हल करना है, प्रश्न-पुस्तिका पर नहीं। यदि छात्र द्वारा 50 से अधिक प्रश्नों को हल किया जाता है तो प्रारम्भिक हल किये हुए 50 उत्तरों को ही मूल्यांकन हेतु सम्मिलित किया जाएगा। सभी प्रश्नों के अंक समान हैं।
- उ. प्रश्नों के उत्तर अंकित करने से पूर्व प्रश्न-पुस्तिका तथा OMR आन्सर-शीट को सावधानीपूर्वक देख लें। दोषपूर्ण प्रश्न-पुस्तिका जिसमें कुछ भाग छपने से छूट गए हों या प्रश्न एक से अधिक बार छप गए हों या उसमें किसी अन्य प्रकार की कमी हो, तो उसे तुरन्त बदल लें।

(शेष निर्देश अन्तिम पृष्ठ पर)

(Only for Rough Work)

1.	which of the following equations shows
	the relationship between free energy
	change (ΔG) and the change in entropy
	(ΔS) , under constant temperature and
	pressure ?

- (A) $\Delta G = T\Delta H \Delta S$
- (B) $\Delta G = T\Delta H/\Delta S$
- (C) $\Delta G = \Delta H/T\Delta S$
- (D) $\Delta G = \Delta H T\Delta S$
- 2. What is the site for gluconeogenesis?
 - (A) Liver
 - (B) Blood
 - (C) Muscles
 - (D) Brain
- Law of thermodynamics which states that energy can neither be created nor be destroyed is ______.
 - (A) The second law of thermodynamics
 - (B) Third law of thermodynamics
 - (C) First law of thermodynamics
 - (D) Zero-order kinetics

- 4. Gibbs' free energy is the portion of the total energy which is available for useful work.
 - (A) True
 - (B) False
- 5. Name the enzyme which is responsible for the conversion of pyruvate to phosphoenolpyruvate (PEP) ?
 - (A) Pyruvate carboxylase
 - (B) Pyruvate carboxykinase
 - (C) Glucose-6-phosphatase
 - (D) Phosphofructokinase
- 6. Which of the following are major sites for glycogen storage?
 - (A) Adipose tissue
 - (B) Bones
 - (C) Muscle and liver
 - (D) Kidney and liver
- 7. What is the value of ΔG , when a system is in equilibrium ?
 - (A) $\Delta G = 0$
 - (B) $\Delta G = 1$
 - (C) $\Delta G = -1$
 - (D) $\Delta G = \Delta G^0$

0	W/L:-16 (1 6-11	10	Name dia amount of the formation of the
8.	Which of the following act as a storage	12.	Name the enzyme which is used for
	form of high energy phosphate ?		branching of glycogen:
	(A) Glucose-6-phosphate		(A) Branching enzyme
	(B) Phosphoenolpyruvate		(B) Hexokinase
	(C) Phosphazens		(C) Phosphoglucomutase
	(D) Glycerol phosphate		(D) Glycogen synthase
9.	Which of the following is the precursor	13.	What is the final product of electron
	of glycogen ?		transport chain ?
	(A) Glycerol-3-phosphate		(A) AMP
	(B) Malate		(B) H ₂ O
	(C) UDP-glucose		(C) ATP
	(D) Leucine and lysine		(D) ADP
10.	What is the name of the molecule which	14.	Glycolysis is also known as
	donates its electrons?		(A) Citric acid cycle
	(A) Reducing agent		(B) Tricarboxylic acid cycle
	(B) Oxidative agent		(C) Kreb's cycle
	(C) Standard reduction potential		(D) Embden-Meyerhof pathway
	(D) Oxidant		
		15.	Citric acid cycle essentially involves the
11.	Which of the following hormones		oxidation of acetyl-CoA to CO ₂
	maintain blood glucose level by		and
	activation of gluconeogenesis?		(A) H ₂ O
	(A) Norepinephrine		-
	(B) Glucagon		(B) O_2
	(C) Insulin		(C) H_2O_2
	(D) Epinephrine		(D) S ₂

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	(D)	None of the above			production of FADH ₂
	(C)	Citrate synthase		(D)	involve loss of CO ₂ and the
	(B)	Pyruvate kinase			production of NADH
	(A)	Hexokinase		(C)	involve loss of CO ₂ and the
	invo	lved in glycolysis metabolism?			production of NAD
18.	Whi	ch of the following enzymes are not		(B)	involve loss of CO ₂ and the
	(D)	Muscle glycogenolysis		` '	do not occur in the TCA cycle
		substances	21		dative decarboxylations:
	(C)	Conversion of non-sugar	21	0.	datina da saukarreletiene
	(B)	Hepatic glycogenolysis		(D)	photophosphorylation
	(A)	Absorbed after digesting food		(C)	glycolysis
	inclu	de		(B)	oxidative phosphorylation
	and	the source of blood sugar does not		(A)	substrate-level phosphorylation
	norm	nal blood is stable at 80-120 mg/dL		resp	iration is generated by :
17.	The	concentration of $C_6H_{12}O_6$ in	20	. Mos	st of the ATP made during cellular
	(D)	carbon dioxide and water		(D)	Acetyl-CoA
	(C)	glucose		(C)	Acetic acid
	(B)	maltose		(B)	Lactic acid
	(A)	starch		(A)	Acetaldehyde
	main	ıly		beer	n converted to:
	eater	n potato in the human body is		ente	rs the citric acid cycle after it has
16.	The	final product of the catabolism of the	19	. Pyrı	evate, the end product of glycolysis,

22.	Which metabolic pathway or process is	25.	The cytochrome c oxidase complex :
	common to both aerobic and anaerobic		(A) accepts electrons from cyt-c.
	oxidation of sugar ?		(B) donates four electrons to O_2 .
	(A) Kreb's cycle		(C) produces 2 H ₂ O per O ₂ reduced
	(B) Chemiosmosis in mitochondrion		(D) All of the above are correct.
	(C) Glycolysis	2.5	
	(D) Oxidation of NAD ⁺ by the electron	26.	Which of the following steps is common in the glycolysis and pentose-phosphate
	transport chain		pathway?
23.	The FADH ₂ and NADH produced by		(A) Conversion of glucose to glucose-6-P
	the oxidation of one acetyl-CoA result in the synthesis of about ATPs.		(B) Conversion of glucose-6-p to ribose-5-P
	(A) 4		(C) Conversion of glucose-6-Pto
	(B) 7		fructose-6-P
	(C) 11		(D) Conversion of glucose to
	(D) 15		glucose-1-P
24.	During electron transport, protons	27.	Pentose-phosphate pathway is
	are pumped out of the mitochondrion		responsible for generating NADPH
	at each of the major sites except		(reducing equivalents in the cell) in the cell. Which of the following enzyme is
	for:		involved in generating NADPH?
	(A) Complex I		(A) Glucose-6-P oxidase
	(B) Complex II		(B) Glucose-6-P dehydrogenase
	(C) Complex III		(C) Glucose-6-P reductase
	(D) Complex IV		(D) Glucose-6-P synthetase

28.	Glucose-6-Phosphate	dehydrogenase	31.	Each cycle of β -oxidation produces :
	is allosterically	inhibited		(A) 1 FAD, 1 NADH and 1 acetyl-CoA
	by			(B) 1 FADH ₂ , 1 NADH and 1 acetyl-
	(A) Acetyl-CoA			CoA
	(B) Citrate			(C) 1 FAD, 1 NAD $^+$, and 2CO $_2$ molecules.
	(C) Glucose			(D) 1 FADH ₂ , 1 NADH and 2CO ₂
	(D) Fructose			molecules.
29.	Bile acid is derived from:		32.	Oxidation of palmitic acid (C16)
	(A) Cholesterol			involves rounds of B-oxidation
	(B) Amino acids			and yields molecules of acetyl-CoA.
	(C) Fatty acids			(A) 8, 8
	(D) Bilirubin			(A) 8, 8 (B) 7, 8
30.	Which of the following i	nhibits acetyl-		(C) 16, 8
	CoA carboxylase-a rate-lin	miting enzyme		(D) 7, 7
	of fatty metabolism ?		33.	Which of the following would yield the
	(A) Citrate			most energy per gram when oxidized?
	(B) ATP			(A) Starch
	(C) Malonyl-CoA			(B) Glycogen(C) Fat
	(D) Acyl-CoA			(D) Protein

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- 34. The acyl-CoA formed in the cytosol is transported to the–for oxidation using a shuttle involving the intermediate formation of acyl _____.
 - (A) mitochondrial matrix, carnitine
 - (B) mitochondrial matrix, coenzyme A
 - (C) endoplasmic reticulum, albumin
 - (D) endoplasmic reticulum, carnitine
- 35. The major difference between the anabolism and catabolism of fatty acids is that:
 - (A) Acetyl-CoA is not required for anabolism.
 - (B) Anabolism occurs in the intermembrane space of mitochondria while catabolism takes place in the mitochondrial matrix.
 - (C) Biotin is required for oxidation of fatty acids but not for anabolism.
 - (D) NADPH is required for anabolismof fatty acids and not for catabolism.

- 36. Ketone bodies are formed when:
 - (A) An organism consumes excessive amounts of carbohydrate.
 - (B) Oxaloacetate is converted to acetyl-CoA.
 - (C) There is a deficiency of acetyl-CoA.
 - (D) There is not enough oxaloacetate to react with available acetyl-CoA.
- 37. Most of the reducing equivalents (NADPH) utilized for the synthesis of fatty acids can be generated from:
 - (A) Glycolysis
 - (B) Mitochondrial malate dehydrogenase
 - (C) The citric acid cycle
 - (D) The pentose-phosphate pathway
- 38. During each cycle of ongoing fatty acid oxidation, all the following compounds are generated, except:
 - (A) Acetyl-CoA
 - (B) Fatty acyl-CoA
 - (C) NADH
 - (D) Water

39.	High content of triglyceride is seen in:	43.	The diet of a child suffering from Maple
	(A) HDL		syrup urine disease (an amino acid
	(B) LDL		disorder), should be low, in which
	(C) VLDL		out of the following amino acids
	(D) Chylomicrons		content ?
40.	Bile acid is synthesized in		(A) Branched chain amino acids
	(A) Kidney		(B) Phenylalanine alanine
	(B) Liver		(C) Methionine
	(C) Pancreas		
	(D) Stomach		(D) Tryptophan
41.	Which out of the following amino acids	44.	Urea is synthesized in:
	is not converted to Succinyl co A?		(A) Cytoplasm
	(A) Methionine		(B) Mitochondria
	(B) Valine		(C) Both cytoplasm and mitochondria
	(C) Isoleucine		(D) Lysosomes
	(D) Histidine	45.	All of the following amino acids are
42.	All of the following compounds are		donors of one carbon compounds,
	synthesized by transmethylation		except:
	reactions, except:		(A) Histidine
	(A) Choline		
	(B) Epinephrine		(B) Tyrosine
	(C) Creatine		(C) Tryptophan
	(D) Ethanolamine		(D) Serine

46.	The two nitrogen of urea are derived	50.	In which form the nitrogen is
	from:		incorporated into an amino acid?
	(A) Aspartate and ammonia		(A) Nitrite
	(B) Glutamate and ammonia(C) Argino succinate and ammonia(D) Alanine and ammonia		(B) Glutamate(C) Nitrate(D) Ammonium ion
47.	Which coenzyme out of the following is required for the oxidative deamination of most of amino acids? (A) Folic acid (B) Pyridoxal-P (C) FMN (D) FAD	51.	An amino group donated by glutamine is attached at C-1 of PRPP, this results in (A) 5-phosphoribosylamine (B) 4-phosphoribosylamine (C) 3-phosphoribosylamine (D) 2-phosphoribosylamine
48.	Choose the incorrect statement about amino acid glycine: (A) One carbon donor (B) Required for the synthesis of haem (C) Forms oxalates upon catabolism (D) Both glucogenic as well as ketogenic	52.	The first intermediate with a complete purine ring is (A) Inosinate (B) Formate (C) Aspartate (D) Glycine
49.	Which out of the following is required as a coenzyme for the transamination reactions? (A) Coenzyme-A (B) Pyridoxal-P (C) Folic acid (D) Cobalamine	53.	Which of the following is an important precursor in the purine pathway? (A) Glycine (B) Aspartate (C) Glutamine (D) Leucine

54.	In the first committed step of pyrimidine	57.	Conversion of dUMP to dTMP is
	biosynthesis, the reaction is catalyzed		catalyzed by
	by		(A) Thymidylate synthase
	(A) Adenylate kinase		(B) Dihydrofolate reductase
	(A) Adenytate kinase		(C) Dihyhroorotase
	(B) Aspartate transcarbamoylase		(D) Cytidylate synthase
	(C) Dihyhroorotase	58.	Adenosine deaminase deaminates
	(D) Cytidylate synthase		adenosine to
			(A) Hypoxanthine
55.	Phosphorylation of AMP to ADP is		(B) Inosine
	promoted by		(C) Xanthine
	(A) Adenylate kinase		(D) Guanosine
	(B) Aspartate transcarbamoylase	59.	Which of the following is degraded to
	•		methylmalonyl semialdehyde?
	(C) Dihyhroorotase		(A) Glutamine
	(D) Cytidylate synthase		(B) Tyrosine
56	CTD is formed from LITD by the action		(C) Thymine
56.	CTP is formed from UTP by the action		(D) Leucine
	of	60.	Which of the following is an important
	(A) Adenylate kinase		precursor in pyrimidine pathway?
	(B) Aspartate transcarbamoylase		(A) Glycine
	(C) Dihyhroorotase		(B) Aspartate
	(C) Diffilloofotase		(C) Glutamine
	(D) Cytidylate synthase		(D) Leucine

(11)

Set-A

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4. Four alternative answers are mentioned for each question as—A, B, C & D in the booklet. The candidate has to choose the most correct/appropriate answer and mark the same in the OMR Answer-Sheet as per the direction:

Example:

Question:

Q. 1 (A) (C) (D) (Q. 2 (A) (B) (D) (D)

Q.3 $\stackrel{\frown}{(A)}$ $\stackrel{\frown}{(C)}$ $\stackrel{\frown}{(C)}$

Illegible answers with cutting and over-writing or half filled circle will be cancelled.

- 5. Each question carries equal marks. Marks will be awarded according to the number of correct answers you have.
- 6. All answers are to be given on OMR Answer sheet only. Answers given anywhere other than the place specified in the answer sheet will not be considered valid.
- 7. Before writing anything on the OMR Answer Sheet, all the instructions given in it should be read carefully.
- 8. After the completion of the examination candidates should leave the examination hall only after providing their OMR Answer Sheet to the invigilator. Candidate can carry their Question Booklet.
- 9. There will be no negative marking.
- 10. Rough work, if any, should be done on the blank pages provided for the purpose in the booklet.
- 11. To bring and use of log-book, calculator, pager and cellular phone in examination hall is prohibited.
- 12. In case of any difference found in English and Hindi version of the question, the English version of the question will be held authentic.
- Impt.: On opening the question booklet, first check that all the pages of the question booklet are printed properly. If there is ny discrepancy in the question Booklet, then after showing it to the invigilator, get another question Booklet of the same series.

4. प्रश्न-पुस्तिका में प्रत्येक प्रश्न के चार सम्भावित उत्तर—
A, B, C एवं D हैं। परीक्षार्थी को उन चारों विकल्पों में से
एक सबसे सही अथवा सबसे उपयुक्त उत्तर छाँटना है।
उत्तर को OMR आन्सर-शीट में सम्बन्धित प्रश्न संख्या में
निम्न प्रकार भरना है:

उदाहरण :

प्रश्न :

प्रश्न 1 (A) (C) (D) प्रश्न 2 (A) (B) (D) प्रश्न 3 (A) (C) (D)

अपठनीय उत्तर या ऐसे उत्तर जिन्हें काटा या बदला गया है, या गोले में आधा भरकर दिया गया, उन्हें निरस्त कर दिया जाएगा।

- 5. प्रत्येक प्रश्न के अंक समान हैं। आपके जितने उत्तर सही होंगे, उन्हीं के अनुसार अंक प्रदान किये जायेंगे।
- 6. सभी उत्तर केवल ओ. एम. आर. उत्तर-पत्रक (OMR Answer Sheet) पर ही दिये जाने हैं। उत्तर-पत्रक में निर्धारित स्थान के अलावा अन्यत्र कहीं पर दिया गया उत्तर मान्य नहीं होगा।
- ओ. एम. आर. उत्तर-पत्रक (OMR Answer Sheet) पर कुछ भी लिखने से पूर्व उसमें दिये गये सभी अनुदेशों को सावधानीपूर्वक पढ़ लिया जाये।
- 8. परीक्षा समाप्ति के उपरान्त परीक्षार्थी कक्ष निरीक्षक को अपनी OMR Answer Sheet उपलब्ध कराने के बाद ही परीक्षा कक्ष से प्रस्थान करें। परीक्षार्थी अपने साथ प्रश्न-पुस्तिका ले जा सकते हैं।
- 9. निगेटिव मार्किंग नहीं है।
- 10. कोई भी रफ कार्य, प्रश्न-पुस्तिका के अन्त में, रफ-कार्य के लिए दिए खाली पेज पर ही किया जाना चाहिए।
- 11. परीक्षा-कक्ष में लॉग-बुक, कैलकुलेटर, पेजर तथा सेल्युलर फोन ले जाना तथा उसका उपयोग करना वर्जित है।
- 12. प्रश्न के हिन्दी एवं अंग्रेजी रूपान्तरण में भिन्नता होने की दशा में प्रश्न का अंग्रेजी रूपान्तरण ही मान्य होगा।

महत्वपूर्ण : प्रश्नपुस्तिका खोलने पर प्रथमतः जाँच कर देख लें कि प्रश्न-पुस्तिका के सभी पृष्ठ भलीभाँति छपे हुए हैं। यदि प्रश्नपुस्तिका में कोई कमी हो, तो कक्षनिरीक्षक को दिखाकर उसी सिरीज की दूसरी प्रश्न-पुस्तिका प्राप्त कर लें।